PHY 5524: Statistical Mechanics

February  $1^{st}$ , 2012 Assignment # 5

(We will solve these problems next Monday, February  $6^{th}$ , in preparation for the first Midterm Exam, to be held on Wednesday, February  $8^{th}$ , 2012.)

## Problem 1

Consider a *classical* system of N non-interacting diatomic molecules enclosed in a box of volume V at temperature T. The Hamiltonian for a *single* molecule is,

$$H(\mathbf{p}_1, \mathbf{p}_2, \mathbf{r}_1, \mathbf{r}_2) = \frac{1}{2m} (\mathbf{p}_1^2 + \mathbf{p}_2^2) + \frac{1}{2} K |\mathbf{r}_1 - \mathbf{r}_2|^2 ,$$

where  $\mathbf{p}_1$ ,  $\mathbf{p}_2$ ,  $\mathbf{r}_1$ , and  $\mathbf{r}_2$  are the momenta and coordinates of the two atoms in a molecule.

- **1.a)** Find the partition function and the Helmholtz free energy of the system (*Hint*: the integration over phase space is easier if, in the integration over  $d\mathbf{r}_1 d\mathbf{r}_2 (= d^3 r_1 d^3 r_2)$  one changes variables from the coordinates of the individual atoms,  $(\mathbf{r}_1, \mathbf{r}_2)$ , to the coordinates of the reduced system,  $(\mathbf{R}, \mathbf{r})$ , where  $\mathbf{R} = (\mathbf{r}_1 + \mathbf{r}_2)/2$  is the position vector of the center of mass of the two atoms, and  $\mathbf{r} = (\mathbf{r}_1 \mathbf{r}_2)$  is the relative distance between the two. In other words, apply what you know about the classical mechanics solution of a two-body problem!).
- **1.b)** Find the specific heat at constant volume.
- **1.c)** Find the mean square molecular diameter,  $\langle |\mathbf{r}_1 \mathbf{r}_2|^2 \rangle$ .

## Problem 2

A simple anti-ferromagnetic system consists of a chain of N non-interacting and identical localized spin-1/2 *dimers*. The system is placed in an external magnetic field  $\mathbf{B} = B\hat{\mathbf{Z}}$  and is in contact with a heat reservoir at a fixed temperature T. The Hamiltonian for an individual dimer is given by the following expression,

$$H = J\mathbf{S}(1) \cdot \mathbf{S}(2) - \mu B \left( S_z(1) + S_z(2) \right)$$

where J and  $\mu$  are positive constants.

- **2.a)** Obtain the eigenvalues and the eigenvectors of H.
- **2.b)** Determine the partition function for the system (i.e. N non-interacting and localized dimers) correct to second order in B, but correct to all orders in T.
- **2.c)** Evaluate the expectation value of the magnetization  $M = \mu \langle S_z(1) + S_z(2) \rangle$  to leading order in *B*. Justify your result in the limit  $T \to 0$ .

2.d) Obtain the magnetic susceptibility defined as follows,

$$\chi = \left(\frac{\partial \langle M \rangle}{\partial B}\right)_{T,B=0}$$

You are told that the magnetic susceptibility is proportional to the variance in the magnetization. Is your answer consistent with this statement? Explain.

## Problem 3

Consider a diluted gas of diatomic molecules. For most such systems, at temperatures above the normal boiling point of the fluid, the rotational energies are so closely spaced that they approximate a continuum, and *classical* statistical mechanics can be used. Let the mass of a molecule be M and the moment of inertia I.

**3.a)** Express the rotational kinetic energy in terms of the two Euler angles  $\phi$  and  $\theta$ , and show that in terms of the corresponding canonical momenta  $P_{\phi}$  and  $P_{\theta}$ , it takes the form,

$$K_{\rm rot} = \frac{1}{2I} \left( P_{\theta}^2 + \frac{P_{\phi}^2}{\sin^2 \theta} \right)$$

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(*Hint*: as in Problem 1, here too you are dealing with a diatomic molecule, i.e. a two-body system. Consider the suggestions given in **Problem 1**).

- **3.b)** Ignoring the vibrations, write down the partition function.
- 3.c) Evaluate the partition function and show that the result is of the form,

$$Z = \frac{V^N}{N!} \left(\frac{2\pi M k_B T}{h^2}\right)^{3N/2} \left(\frac{8\pi^2 I k_B T}{h^2}\right)^N$$

- **3.d)** Compute the internal energy and the entropy of this gas.
- **3.e)** Using the expressions from (**3.d**), examine the behavior of the entropy at low temperatures. Does the obtained result agree with the third law of thermodynamics? Explain what happens at low temperatures.